

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance  
 (B) volume, molarity, pH, amount of substance  
 (C) volume, molarity, pH, concentration of ions  
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A)  $n = 1000$ ,  $m = 1000$

(B)  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$ ,  $10 \text{ g H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$   
 (C)  $1 \text{ mol H}_2\text{O}$ ,  $1 \text{ mol H}_2\text{O}$

2. In a given solution, the concentration of a weak acid is  $10^{-2} \text{ mol dm}^{-3}$  and the concentration of its conjugate base is  $10^{-3} \text{ mol dm}^{-3}$ . What is the pH of the solution? (correct)

(A)  $\text{pH} = 2.0$ ,  $\text{pH} = 3.0$ ,  $\text{pH} = 4.0$ ,  $\text{pH} = 5.0$

(B)  $10^{-2} \text{ mol dm}^{-3}$ ,  $10^{-3} \text{ mol dm}^{-3}$ ,  $10^{-2} \text{ mol dm}^{-3}$ ,  $10^{-3} \text{ mol dm}^{-3}$ ,  $10^{-2} \text{ mol dm}^{-3}$ ,  $10^{-3} \text{ mol dm}^{-3}$

3. The pH of a solution of  $\text{CH}_3\text{COOH}$  is 4.0. What is the concentration of  $\text{CH}_3\text{COO}^-$  ions in the solution? (correct)

(A)  $10^{-4} \text{ mol dm}^{-3}$ ,  $10^{-3} \text{ mol dm}^{-3}$ ,  $10^{-2} \text{ mol dm}^{-3}$ ,  $10^{-1} \text{ mol dm}^{-3}$

4. Which of the following is a weak acid? (correct)



(B)  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{O}$

5. A solution of a weak acid is  $0.1 \text{ mol dm}^{-3}$  and its pH is 3.0. What is the concentration of the conjugate base? (correct)



(B)  $10^{-3} \text{ mol dm}^{-3}$ ,  $10^{-2} \text{ mol dm}^{-3}$ ,  $10^{-1} \text{ mol dm}^{-3}$ ,  $10^{-4} \text{ mol dm}^{-3}$

6. In a weak acid solution, the concentration of the conjugate base is  $10^{-3} \text{ mol dm}^{-3}$  and the concentration of the weak acid is  $10^{-2} \text{ mol dm}^{-3}$ . What is the pH of the solution? (correct)



# Multiple Choice Stoichiometry Test

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