

Unit 1 Study Guide — ANSWER KEY

1. Define scientific method: **a procedure for collecting, testing, and communicating information**
2. Define hypothesis: **a suggested answer to the question, an educated guess**
3. Define experiment: **a scientific test so see what will happen**
4. Define dependent variable: **the variable that you do NOT control. (on a graph is usually on the Y axis)**
5. Define independent variable: **the variable that I control. (on a graph is usually on the X axis)**
6. Define control: **the part of the experiment where everything is left the same.**
7. Define observation: **information collected with any of your 5 senses**
8. Define density: **the ratio of mass per unit volume (D = m/v)**
9. Define accuracy: **the measure of how close your data is to the ACTUAL or true value.**
10. Define precision: **a measure of how close your data (measurements) are to each other.**
11. Why do scientists use SI units of measurement? **So that measurements can be understood internationally (around the world).**

12. List (in the correct order) the steps of the scientific method.

1. make an observation	4. test hypothesis (experiment)
2. ask a question	5. analyze data/form a conclusion
3. form a hypothesis	6. develop a theory

13. What mnemonic device is used to remember the metric system?

K H Da B d c m
g, m, L.

14. What is the sentence (mnemonic) that will help you remember the METRIC prefixes in order?

King Henry Died By Drinking Chocolate Milk

15. Fill in the table below describing the metric system:

	Kilo	Hecto	Deca	Base	Deci	Centi	Milli
Abbreviation	K	H	Da	g, m, L	d	c	m
Meaning in powers of 10	10^3	10^2	10^1	10^0	10^{-1}	10^{-2}	10^{-3}

Convert the following:

- | | |
|---------------------------------|------------------------------------|
| 16. 25L to <u>25000</u> mL | 21. 25 cm to <u>0.25</u> m |
| 17. 2.3 km to <u>230,000</u> cm | 22. 56784 mm to <u>0.056784</u> km |
| 18. 0.045 g to <u>45</u> mg | 23. 15 g to <u>0.015</u> kg |
| 19. 0.35 kg to <u>350</u> g | 24. 3.452 kL to <u>3452</u> L |
| 20. 300 mL to <u>0.300</u> L | 25. 47 m to <u>47000</u> mm |

Scientific Notation:

Convert the following to scientific notation:

- | | | | |
|------------|--------------------------------------|------------|--|
| 26) 0.005 | <u>5×10^{-3}</u> | 30) 0.025 | <u>2.5×10^{-2}</u> |
| 27) 5.050 | <u>5.05×10^0</u> | 31) 0.0025 | <u>2.5×10^{-3}</u> |
| 28) 0.0008 | <u>8×10^{-4}</u> | 32) 500 | <u>5×10^2</u> |
| 29) 1,000 | <u>1×10^3</u> | 33) 5,000 | <u>5×10^3</u> |

700 Answer Key Study Guide

J Elliott



700 Answer Key Study Guide:

Unveiling the Magic of Words: A Review of "700 Answer Key Study Guide"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "700 Answer Key Study Guide," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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