

## 14.1

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and of species associated with them.
2. Calculate hydroxide ion concentrations, molar ratios, and pH values for weak acid-conjugate base systems.
3. Calculate the pH of weak acid solutions.

## Vocabulary

- |                   |                             |                  |
|-------------------|-----------------------------|------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. acid solution |
| 4. hydroxide ion  | 5. molar ratio              | 6. hydroxide ion |
|                   | 7. weak acid                | 8. pH            |

## Key Equations

- |  |   |
|--|---|
| 1. $\text{pH} = -\log[\text{H}^+]$ (Eq. 14.1)        | 2. $\text{pOH} = -\log[\text{OH}^-]$ (Eq. 14.2) |
| 3. $\text{pH} + \text{pOH} = \text{pK}_w$ (Eq. 14.3) | 4. $\text{pH} + \text{pOH} = 14.00$ (Eq. 14.4)  |

## Part A: Completion

Use the appropriate number of significant figures in the answers to the questions that are indicated in the number of significant figures indicated in the number of places.

1. What products are **2** formed when hydrochloric acid ( $\text{HCl}$ ) reacts with  $\text{NaOH}$ ? Write the balanced equation for the reaction and identify the products as strong electrolyte, **2** weak electrolyte, or none.
2. Hydrochloric acid has a molar mass of **36.5** g/mol. How many **2** moles of hydrochloric acid are present in **36.5** g of hydrochloric acid?
3. How many **2** moles of hydrochloric acid are present in **36.5** g of hydrochloric acid?
4. How many **2** moles of hydrochloric acid are present in **36.5** g of hydrochloric acid?
5. How many **2** moles of hydrochloric acid are present in **36.5** g of hydrochloric acid?
6. How many **2** moles of hydrochloric acid are present in **36.5** g of hydrochloric acid?
7. How many **2** moles of hydrochloric acid are present in **36.5** g of hydrochloric acid?

8. **0.100**
9. **0.100**
10. **0.100**
11. **0.100**
12. **0.100**
13. **0.100**
14. **0.100**
15. **0.100**
16. **0.100**
17. **0.100**
18. **0.100**

# 19 2 Hydrogen Ions And Acidity Answer Key

**RD Boyd**



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