

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are the characteristic properties for a weak acid? _____
- (A) pKa and molar conductivity (B) pKa and pH
- (C) molar conductivity (D) molar conductivity and pH

2. Which of the following is CORRECT? In a weak acid solution, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L
- (C) 10^{-3} mol/L (D) 10^{-2} mol/L
3. In a 0.1 M solution of a weak acid, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L

4. In a 0.1 M solution of a weak acid, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L
5. In a 0.1 M solution of a weak acid, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L

6. In a 0.1 M solution of a weak acid, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L

7. Which of the following is CORRECT? In a weak acid solution, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L

8. In a 0.1 M solution of a weak acid, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L

9. In a 0.1 M solution of a weak acid, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L

10. In a 0.1 M solution of a weak acid, the concentration of the undissociated acid is _____
- (A) 10^{-4} mol/L (B) 10^{-5} mol/L (C) 10^{-3} mol/L (D) 10^{-2} mol/L



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