

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) 10^{-14} (B) 10^{-7}

2. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

3. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

4. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

5. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

6. Which of the following is correct? (correct)

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

7. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

8. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

9. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

10. In a 0.1 mol/L acetic acid solution, which of the following is correct? (correct)
- (A) $[H^+] = 10^{-7}$ mol/L (B) $[H^+] = 10^{-8}$ mol/L (C) $[H^+] = 10^{-9}$ mol/L (D) $[H^+] = 10^{-10}$ mol/L

(A) 10^{-7} (B) 10^{-8} (C) 10^{-9} (D) 10^{-10}

Multiple Choice Stoichiometry Test

Xiang Xie



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lets the teacher know that you are involved in and invested in your child s education

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be sure to encourage your students to write thank you notes as well dear principal gonzalez

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principal use these captivating and unique notes you are trustworthy sincere and devoted to your work you always take things that concern the staff or the students seriously thank you so much principal