

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe the components and direction of electron flow.

- a. anode \_\_\_\_\_
- b. cathode \_\_\_\_\_
- c. anion \_\_\_\_\_
- d. cation \_\_\_\_\_
- e. salt bridge (indicate flow) \_\_\_\_\_
- f. displacement (write balanced) \_\_\_\_\_
- g. write overall balanced equation \_\_\_\_\_
- h.  $E_{cell}$  \_\_\_\_\_
- i. Write overall cell process with phases.  $\text{cell} = \text{anode} | \text{anode solution} || \text{cathode solution} | \text{cathode}$  represent by net cell equation with full balancing for charges & mass

\_\_\_\_\_

# Voltaic Cells Study Guide

**Joan I. Senyk, Darrell D. Ebbing**



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