

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) $n = 1000$, $m = 1000$

(B) $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $1 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$
 (C) $1 \text{ mol H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$

2. An acid-base reaction represents a proton transfer from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.
 (B) The equilibrium constant is less than 1.
 (C) The reaction is exothermic.
 (D) The reaction is endothermic.



(B) $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O} + \text{H}_2\text{O}$, $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O} + \text{H}_2\text{O}$

3. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.
 (B) The equilibrium constant is less than 1.
 (C) The reaction is exothermic.
 (D) The reaction is endothermic.

(A) $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$, $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O} + \text{H}_2\text{O}$

4. Which of the following is not a property of a strong acid? (correct)



(B) $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O} + \text{HCl}$, $\text{HCl} + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{O} + \text{HCl}$

5. A chemical reaction involves the transfer of a proton from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.
 (B) The equilibrium constant is less than 1.
 (C) The reaction is exothermic.
 (D) The reaction is endothermic.



(B) $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $1 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$, $10 \text{ g H}_2\text{O}$, $1 \text{ mol H}_2\text{O}$

6. An acid-base reaction involves the transfer of a proton from a stronger acid to a weaker base. Which of the following is not true? (correct)
- (A) The equilibrium constant is greater than 1.
 (B) The equilibrium constant is less than 1.
 (C) The reaction is exothermic.
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Multiple Choice Stoichiometry Test

Paul Foglino



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