

NMLS Review
Test 6
03/21/2013

1. What is another name for the "1003"?
 - A. Uniform Residential Loan Application
 - B. Good Faith Estimate
 - C. HUD-1
 - D. TIL Disclosure Statement

2. The following instances would require a borrower to obtain flood insurance:
 - A. The borrower feels more secure with the flood insurance coverage
 - B. The lender policy is all conforming loans require flood insurance
 - C. The appraiser determines the property is in a flood zone
 - D. There was flooding history with the previous owner

3. The borrower does not want to answer the questions in the government monitoring section. You must:
 - A. Refuse to take the application
 - B. Leave that section blank
 - C. Tell the borrower it is a federal violation of the Truth in Lending Act
 - D. Complete the section based on your personal observation

4. Which of the following is voluntarily disclosed by the borrower on the application?
 - A. Race and marital status
 - B. Marital status and age
 - C. Race and sex
 - D. Sex and age

5. The Housing Expense ratio is concerned with the following:
 - A. Borrower's level of income in ratio to the loan amount
 - B. The borrower's credit worthiness
 - C. The borrower's ability to make their monthly mortgage payment
 - D. The borrower's ability to meet housing expenses determined by their income

6. Per FNMA underwriting guidelines, the appraiser can use net adjustments up to _____% on a residential 1-4 family residential appraisal?
 - A. 25%
 - B. 20%
 - C. 15%
 - D. 10%

7. A borrower applies for a refinance loan on their property appraised at \$235,000. There is a open HELOC with a \$40,000 credit limit and the borrower owes \$25,000. The borrower applies for a first mortgage loan amount of \$166,000. What is the LTV, CLTV and TLTV based on the information given.
 - A. 70% / 80% / 95%
 - B. 70% / 83% / 0100%
 - C. 71% / 81% / 98%
 - D. 75% / 80% / 91%

8. A borrower refinances their mortgage and is required to obtain PMI on first mortgages with a LTV over 80%. The appraised value is \$221,000 and the borrower carries a HELOC balance of \$46,000. The borrower applies for a fixed rate mortgage of \$175,000 at 6.5% for 15 years. The following would apply:
 - A. The LTV is 100%
 - B. The borrower is required to have PMI
 - C. The lender will not require the borrower to have PMI on this loan
 - D. The LTV is 90%

Quizes For Uniform State Test Study Guide

CL Gary



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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids e g sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the K_a or K_b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web jul 19 2023 at the beginning of the titration the solution in the erlenmeyer flask is acidic as the base is added it completely reacts with the acid and the solution in the erlenmeyer flask continues to be acidic but at the equivalence point

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral ph at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then $\text{pH} = \log 0.100 = 1.000$ pH log 0.100 = 1.000 b titrant volume 12.50 ml

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