

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. Anode: _____
 - b. Cathode: _____
 - c. Anion: _____
 - d. Cation: _____
 - e. Half-reaction at anode: _____
 - f. Half-reaction at cathode: _____
 - g. Net redox reaction: _____
 - h. Salt: _____
 - i. Write overall cell equation with phases: _____
- Write the spontaneous redox reaction with phases, which is spontaneous, and the cell notation, separated by a double vertical line, with the anode on the left and the cathode on the right. (5 marks)

Voltaic Cells Study Guide

William H. Myers, William T. Mooney



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