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 $2\ 3\ \text{Ca(OH)}_2 + 1\ \text{Al}_2\text{SO}_4 \rightarrow 3\ \text{CaSO}_4 + 2\ \text{Al(OH)}_3$ type of reaction double displacement
 $3\ 3\ \text{Mg} + 1\ \text{Fe}_2\text{O}_3 \rightarrow 2\ \text{Fe} + 3\ \text{MgO}$ type of reaction single displacement
 $4\ 1\ \text{C}_2\text{H}_4 + 3\ \text{O}_2 \rightarrow 2$

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web to the left of the arrow we have the before situation this side represents the substances we have before the reaction takes place they are called the reactants to the right of the arrow we have the after situation this side represents the substances that we have after the reaction has taken place they are called the products

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 $1\ 1\ \text{CaF}_2 + 2\ \text{Na} \rightarrow 2\ \text{NaF} + 1\ \text{Ca}$ type of chemical reaction single displacement
 $2\ 1\ \text{CaCO}_3 \rightarrow 1\ \text{CaO} + 1\ \text{CO}_2$ type of chemical reaction decomposition

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